Electrical and Optical Enhancement Properties of Metal/Semimetal Nanostructures for Metal Oxide UV Photodetectors

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Introduction

UV photodetectors have been investigated for various commercial and military applications, such as secure space-to-space communications, pollution monitoring, water sterilization, flame sensing, and early missile plume detection [1]. To date, epitaxially grown or bulk wide bandgap semiconductors such as GaN, AIN, AlGaN, C (diamond), and SiC have been used for ultraviolet detection [2–12]. Fabrication of devices from these materials is often expensive with intricate processes. Also organic semiconductor materials are an attractive alternative [13, 14], though they lack the carrier mobility of inorganic semiconductors. Metal oxide semiconductor nanomaterials have the advantages of low processing cost, ease of fabrication, a large surface-to-volume ratio with carrier and photon confinement, and amenability to surface functionalization for hybrid inorganic–organic configurations. Furthermore, the unique combination of the carrier transport mechanism and oxygen adsorption/desorption processes on the nanostructures surface leads to a high internal gain. However, they also lead to slow transient response with response time on the order of seconds.

Significant enhancement of photodetector performance can be enabled by layers of metal/semimetal nanostructures. The trade-off between sensitivity and time response may be significantly reduced with careful design of metallic nanostructures relative to metal oxide active regions. The primary mechanisms for enhancement include surface plasmon resonance and carrier transfer to highly conductive materials. Details of the fundamental concepts, parameters of influence, and comparisons of enhanced devices are presented.

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Metal Oxide UV Photodetector

Materials

Metal oxides have been developed with a range of quantum confinement options from bulk, single crystals to 0D quantum dots. Much attention has been placed on nanoscale dimensions. For ultraviolet detection, ZnO has led the way as the most prominent semiconductor among the metal oxides, though In₂O₃, SnO₂, Ga₂O₃, WO₃, and CeO₂ are all viable materials for either solar-blind or visible-blind applications. Table 36.1 shows their basic properties.

Detection Mechanism

The mechanism for ultraviolet detection and subsequent carrier transport for metal oxide nanostructures is the cause of the high photocurrent-to-dark-current ratio characteristic of these materials. A significant change of conductance upon the UV illumination can be described by the following semiquantitative model as described in Wu et al. [15] (Fig. 36.1):

$$G = G_0 \, \exp\!\left(-\frac{q\psi_s}{kT}\right)$$

The activation energy, $q\psi_s$, is located in the interspace between nanocrystals, k is the Boltzmann constant, and T is the absolute temperature. Electrons must overcome the barrier $q\psi_s$; thus, the conductance is exponentially dependent on the barrier height. In the dark, due to the affinity between the oxygen molecules and electrons, oxygen molecules adsorb onto metal oxide surface and capture nearby electrons to form negatively charged oxygen ion layer. This process can be represented by

$$O_2(g) + e^- \rightarrow O_2^-(ad)$$

As a result, a depletion layer is formed near the surface of metal oxide and the barrier height increases, which results in reduced carrier concentration and low conductance. For low-dimensional metal oxide materials, this effect can be extremely strong due to the high surface-to-volume ratio, and the depletion region created in the oxygen adsorption process can extend throughout the entire film. Upon exposure to light with photon energy above the bandgap of the metal oxide, electron–hole pairs are generated.

$$hv \rightarrow e^- + h^+$$

The photogenerated holes migrate to the surface of metal oxide and discharge the negatively charged adsorbed oxygen ions through surface electron-hole recombination.

Metal oxides	Crystal structure	Bandgap (eV)	Spectrum ^a
ZnO	Hexagonal	3.37	UV A
In ₂ O ₃	Cubic/	3.5–3.7	UV A (direct)
	rhombohedral	2.9	Blue (indirect)
SnO ₂	Tetragonal	3.6	UV A
Ga ₂ O ₃	Monoclinic	4.2-4.9	UV B-UV C
WO ₃	Monoclinic	3.3	UV A
CeO ₂	Cubic	3.2	UV A
	Metal oxides ZnO In ₂ O ₃ SnO ₂ Ga ₂ O ₃ WO ₃ CeO ₂	Metal oxides Crystal structure ZnO Hexagonal In ₂ O ₃ Cubic/ rhombohedral SnO ₂ Tetragonal Ga ₂ O ₃ Monoclinic WO ₃ Monoclinic CeO ₂ Cubic	MetalBandgap (eV)znOHexagonal3.37In2O3Cubic/ rhombohedral3.5-3.7 2.9SnO2Tetragonal3.6 Ga2O3Ga2O3Monoclinic4.2-4.9WO3Monoclinic3.3 CeO2

^aUltraviolet A (3.10–3.94 eV, 400–315 nm), ultraviolet B (3.94–4.43 eV, 315–280 nm), ultraviolet C (4.43–12.4 eV, 280–100 nm)



Fig. 36.1 Depletion region and potential barrier for (a) UV off and (b) UV on (Modified from [15])

$$h^+ + \mathrm{O}_2^-(ad) \to \mathrm{O}_2(g)$$

The barrier and depletion width decreases, and thus, the photocurrent exponentially increases until saturation.

The strong influence of surface effects on the high internal gain also results in a slow transient response. The slow decay time can be attributed to the persistent photoconductivity (PPC) phenomenon. Prades et al. discuss the origin and models for PPC in ZnO nanowires specifically through the principle that can be applied to all metal oxide nanostructures [16]. Two opposite models are proposed. The first is due to bulk defects between shallow and deep energy levels, while the second is a pure surface effect with the capture of electrons by surface states. The relative persistence of conductance, G, is defined as $\frac{\Delta G}{G_0}$ where G_0 is the initial value in darkness. For the metal oxide nanowire, the electron-hole pairs created by UV photons give a photoresponse (ΔG_{ph}) with excess carriers (Δn , Δp):

$$\Delta G_{ph} \propto \Delta n = \Delta p = \frac{g}{1/\tau_{bulk} + 1/\tau_{surf}}$$

where g is the photogeneration rate of carriers per unit volume and τ_{bulk} and τ_{surf} are the lifetimes of photocarrier recombination in the bulk and at the surface. When surface states dominate, which is the case with nanostructures, oxygen adsorbs on the surface to separate holes from electrons forming a built-in potential. The recombination rate is significantly reduced and strongly dependent on oxygen content in air.

Photodetector Figures of Merit

Commonly used figures of merit for photodetectors are summarized in Table 36.1. Responsivity and normalized detectivity are both measures of detector sensitivity. Responsivity is essentially a measure of the effectiveness of the detector to take a given light input and convert it to current. The responsivity can be expressed by

$$R = \frac{i_{out}(\lambda, f)}{P_{in}} \quad A/W$$

where i_{out} is the photogenerated current of the detector, measured in amperes (A), and P_{in} is the incident radiation power, measured in watts (W).

Noise-equivalent power (NEP) is another basic indicator of the performance of a photodetector, which is defined as the signal power that gives a signal-to-noise ratio of one in a one hertz output bandwidth. The NEP can be expressed by the following equation:

$$NEP = \frac{i_n}{R}$$

where the i_n is the noise current in the unit of A/Hz^{1/2} and the *R* is the responsivity in the unit of A/W.

The term normalized detectivity (D^*) is proportional to the reciprocal of the NEP and is independent of the detector area. It is therefore a popular measure of the intrinsic merit of a photodetector material:

$$D^* = \frac{\sqrt{A \times \Delta f}}{NEP}$$
 Jones $\left(\text{cm}\sqrt{\text{Hz}}/\text{W} \right)$

where A is the active area in cm⁻² and Δf is the operating bandwidth in hertz (unity: 1Hz). On the assumption that noise current is dominated by shot noise in the dark current $(i_n = i_{shot} = \sqrt{2qi_{dc}\Delta f})$,

$$D^* = R \sqrt{\frac{A}{2qi_{dc}}}$$
 Jones

where i_{dc} is dark current.

The dark current is measured current under dark conditions. When divided by the active area of the device, it becomes dark current density. Often a photocurrentto-dark-current ratio is defined for a given bias.

The quantum efficiency (QE or η) is another measure of the effectiveness of the radiant energy producing electrical current in a detector. It is the ratio of photogenerated carriers and incident photons.

$$\eta = \frac{i_{out}}{q} \times \frac{hv}{P_{in}} (100) \ \%$$

The relationship between η and R is defined by the following equation:

$$R = \frac{\eta \lambda(\mu m)}{1.24} \quad (A/W)$$

Internal gains are defined by the ratio of photogenerated carriers to absorbed photons, while external gains are defined by the ratio of photogenerated carriers to incident photons. Overall the gain of a photodetector defines the number of electron–hole pairs produced from one photon.

The time or transient response of photodetectors is defined in both rise time and fall time. The incident light source is switched on and off at a sufficient frequency to determine the detector's ability to respond to pulsed light. Rise time is measured between 10 % and 90 % as current increases, while fall time is measured between 90 % and 10 % as current decreases.

Detector types for ultraviolet detection using metal oxides include photoconductors (PC), photodiodes (PD), metal-semiconductor-metal (MSM), metal insulator semiconductor (MIS), and field effect transistors (FET). Figure 36.2 shows schematics of device structures.

Performance Comparison

ZnO Photodetectors

ZnO materials possess various attractive characteristics for ultraviolet optoelectronic devices with reported bandgap values ranging from 3.2 to 3.4 eV, a large excitonic binding energy (60 meV), high radiation harness, amenability to wet chemical etching, and a relatively low growth temperature.



Fig. 36.2 Device design examples of ultraviolet photodetectors (Modified from [17–21])

Table 36.2 compares devices for metal oxides using selected figures of merit. Thin films have been developed by various deposition techniques including RF magnetron sputtering [22], molecular beam epitaxy (MBE), metalorganic chemical vapor deposition (MOCVD), pulsed laser deposition (PLD), and the sol-gel process. In 2000 and 2001, Liu et al. developed metal-semiconductor-metal and Schottky UV photodetectors with high-quality, epitaxially grown, ZnO thin films using MOCVD [23, 24]. Yang et al., in 2002, were the first to discover extreme sensitivity to ultraviolet light with ZnO nanowires (1D nanostructures), grown using a vapor phase transport process [25]. In 2007, Soci et al. investigated the inherent internal gain mechanism and consequential slow transient response of ZnO nanostructures [26]. Solution-processed colloidal nanoparticles (0D nanostructures) were developed by Jin et al. in 2008. The existence of persistent photoconduction found in solution-processed hybrid inorganic-organic nanostructures resulting in slower response time was investigated by Li et al. (Table 36.3).

Sawyer et al. explored surface passivation, increasing photocurrent sensitivity, for high-quality ZnO nanocrystals made by top-down wet-chemical etching technique developed by Dutta et al. [36, 38].

Other Metal Oxide UV Photodetectors

Research has expanded detection wavelengths throughout the UV range with materials such as indium oxide (In_2O_3) , tin dioxide (SnO_2) , gallium oxide $(\beta-Ga_2O_3)$, and cerium oxide or ceria (CeO_2) . Though typically the detection

	Symbol/		
Figure-of-Merit	acronym	Unit	Definition
Responsivity	R	A/W	Photocurrent divided by incident optical power
Normalized detectivity	D*	$cm \frac{\sqrt{Hz}}{W}(Jones)$	Measure of detector sensitivity that enables comparison even when detector area and bandwidth are different
Dark current	I_d	А	Current in the absence of light
Dark-current density	J_d	$A \text{ cm}^{-2}$	Current divided by device active area in the absence of light
UV photogenerated current to dark current ratio	UV/dark ratio	Unitless	
External quantum efficiency	EQE or η	%	The ratio of photogenerated carriers to photon incident on the device
Internal gain	G _{int}	Unitless	Ratio of photogenerated carriers to photon absorbed
External gain	G _{ext}	Unitless	Ratio of photogenerated carriers to photon incident $G_{ext} = \eta \times G_{int}$
Noise-equivalent power	NEP	W/\sqrt{Hz}	Minimal detectable power, the optical signal in watts at which the electrical signal of noise ratio is unity when the bandwidth is limited to 1 Hz
Response time or transient response	t _r rise time	S	Amount of time required for current to rise from 10 % to 90 % of maximum current
	t _f fall time		Amount of time required for current to fall from 90 % to 10 % of maximum current

Table 36.2 Photodetector figures of merit

mechanisms are the same, physical properties of each material vary which contribute to differences in device performance.

Tin-doped indium oxide (ITO) is a pervasive material in electronics and photonics for its use as a transparent conducting electrode. However, the electronic and optical properties of undoped indium oxide, In_2O_3 , especially in nanostructure form, have only recently been investigated. In 2001, Liang et al. were among the first to measure the semiconducting properties of In_2O_3 nanofibers grown by goldseeded vapor–liquid–solid (VLS) mechanism [39]. Since then, nanowires, nanotowers, nanopyramids, and colloidal nanoparticles have been characterized for their electronic and optical properties [39–47]. However, the complete photoresponse including responsivity and time response of an In_2O_3 ultraviolet photodetector was only recently investigated [48]. A maximum responsivity of 11 A/W was observed at 340 nm, and the rise and fall times were 1,100 and 3,200 s, respectively.

Tin oxide (SnO_2) materials were historically researched for use as a conductive electrode and antireflective coating due to its transparency to visible light. As a wide bandgap material (3.6 eV), its potential as an ultraviolet detector has been explored with the creation of FETs and photoconductors using SnO₂ nanowire,

						D^*	Time respoi	nse (s)		
			Responsivity	Dark current		(cm, Hz ^{0.5} .			Device	
Ref.	Year	λ (nm)	$(A W^{-1})$	(A)	Area (cm ²)	W^{-1})	Rise time	Fall time	type	Material
[23]	2000	300-373	400	$4.50 imes 10^{-7}$	I	1	1×10^{-6}	$1.5 imes10^{-6}$	MSM	Epitaxial thin film MOCVD
[24]	2000	300-385	1.5	$1 imes 10^{-9}$	$1.77 imes10^{-4}$	1.12×10^{12}	$1.2 imes10^{-8}$	$5 imes 10^{-8}$	MSM	Epitaxial thin film MOCVD
[26]	2007	245-385	$3.33 imes 10^4$	10^{-9}	3×10^{-7}	$2.04 imes10^{13}$	~ 50	>100	PC	Nanowires
[27]	2008	370-395	61	$1.2 imes10^{-10}$	$3.14 imes10^{-2}$	$1.91 imes 10^{15}$	25	120	MSM	Colloidal nanoparticles
[22]	2007	300-384	2,069	$2.22 imes 10^{-6}$	$8.85 imes10^{-3}$	$2.31 imes10^{14}$	I	I	MSM	Thin film grown by RF
										sputtering
[17]	2008	300-400	$8.9 imes10^{-2}$	$4.11 imes 10^{-7}$	$4 imes 10^{-4}$	$4.91 imes 10^9$	I	I	MSM	Epi layer grown by RF
			$8.3 imes 10^{-3}$	$2.22 imes10^{-10}$	$4 imes 10^{-4}$	1.97×10^{10}	I	I	MIS	plasma-assisted MBE (Omni
										Vac) on sapphire (0001) substrates
[28]	2008	300-400	0.18	1.8×10^{-5} (A)	/cm ²)	$7.5 imes10^{10}$	1	I	I	Nanorods
[29]	2008	300-420	0.68	$9.4 imes10^{-7}$ (A	/cm ²)	$1.24 imes 10^{12}$	<0.2	<0.2	PD	Nanoparticles
[30]	2009	325-375	12.5	I	I	1	I	I	I	Nanoparticles
[31]	2009	310-450	$1 imes 10^{-4}$	4×10^{-13}	$1.2 imes10^{-4}$	3.06×10^9	48	0.9	PC	Nanoparticles

Table 36.3 Performance comparison of ZnO photodetectors

[32]	2009	115-400	$3.3 imes 10^{-10}$	$4.1 imes 10^{-9}$	$4 imes 10^{-6}$	18.22	>1,000 ms	I	MSM	Lateral nanowires on a ZnO: Ga/glass template
33	2009	300-370	0.011	$4.32 imes 10^{-9}$	0.24	1.45×10^{11}	I	1	MSM	Nanowires
34	2009	400-600	$1.5 imes10^{-5}$	1	1	1	1	I	PC	ZnO-infiltrating CQD films
[15]	2010		$4.29 imes 10^{-5}$	$2.23 imes10^{-9}$	25	$8.03 imes 10^9$	I	I	PC	Nanowires and nanoparticles
[35]	2010	200-425	0.02287	$2 imes 10^{-9}$ (A/c	m ²)	$9.04 imes 10^{11}$	1	1	PD	Polymer/nanoparticle bilayer
[36]	2010	200-380	0.33	4.88×10^{-12}	$7.85 imes10^{-3}$	$2.34 imes10^{13}$	100	25	PC	PVA-coated colloidal
										nanoparticles
18	2010	200-380	731.42	2.91×10^{-12}	$6.24 imes10^{-3}$	$5.98 imes10^{16}$	0.6	8	MSM	Colloidal nanoparticles
[19]	2013	320–380	$1.19 imes 10^{-4}$	$4.2 imes10^{-13}$	$7.85 imes10^{-3}$	$2.88 imes 10^{10}$	0.025	0.05	PD	Back-illuminated p-GaN/ZnO
										nanoparticle heterojunction diode
[37]	2013	365	6.83×10^{1}	6.83×10^{-9}		1.46×10^{15}	0.034	0.132	PC	Nanorods passivated by TOPO
										10101

nanobelts, and nanonets. These materials were fabricated using the vapor–liquid– solid process [49–51], laser ablation [52], and thermal evaporation [53]. A SnO₂ nanonet photoconductor was fabricated using a colloid crystal polymer template which forms a 2D ordered semiconducting nanofilm with hole sizes that can be adjusted by the mean diameter of the polymer colloidal spheres. Though responsivity characterization for ultraviolet photodetectors has not yet been reported, FET devices demonstrated on/off ratios of 10³ and 10⁵ with nanowire and nanobelts, respectively [52, 53]. High gain values have also been reported for the photoconductor devices up to 1.32×10^7 for SnO₂ nanowires [49, 50].

Solar-blind detection below 290 nm has been demonstrated with β -Ga₂O₃. Its bandgap has been reported with values between 4.2 and 4.9 eV. Photodetectors were fabricated using thermal evaporation and a single-step chemical vapor deposition process. In recent work, a bridged nanowire photoconductor demonstrated a high 250–280 nm rejection ratio of 2 × 10⁴ and a high photocurrent-to-dark-current ratio of 3 × 10⁴ [54]. In comparison to In₂O₃- and ZnO-based photodetectors, the bridged β -Ga₂O₃ NWs photoconductor can achieve much faster transient response with decay times less than 20 ms. The fast transient response is attributed to the extremely low electron density of β -Ga₂O₃ NWs in dark environment, which significantly suppressed the formation of surface states that originate from oxygen adsorption process [55, 56]. Therefore, the PPC phenomenon was greatly reduced in that oxygen adsorption has a nearly negligible effect on conductance.

Ceria or cesium oxide CeO_2 and tungsten oxide WO_3 are lesser-known metal oxides for UV detection. Ceria have reported bandgap values of 3.2 eV and 3.3 eV, respectively. Both were developed as nanowire devices [57–59] and possess similar high gain values and longtime responses as In_2O_3 and ZnO materials (Table 36.4).

Nanostructures for Performance Enhancement

Research to improve sensitivity and transient response of photodetectors is being pursued across the UV, visible, and infrared (IR) wavelength regions. To date, much of the research has focused on wavelengths from the visible to IR region. Until only recently investigations of enhancement for shorter UV wavelengths begun [61]. Mechanisms for performance enhancement can be categorized as either plasmonic interactions with an electric field induced by metal/ semimetal nanostructures or carrier transfer from metal oxide active regions to highly conductive materials. Surface plasmon resonance (SPR) and carrier transfer concepts are described below with examples of nanostructures and performance comparisons.

Surface Plasmon Resonance Mechanisms

Metallic nanostructures absorb light due to the coherent oscillation of conduction band electrons induced by the interacting electromagnetic field. An incident

								Time			
								respons	e (s)		
Responsivity UV-to-dark- Dark	Responsivity UV-to-dark- Dark	Responsivity UV-to-dark- Dark	UV-to-dark- Dark	Dark			D^* (cm.	Rise	Fall		
Year λ (nm) (A W ⁻¹) current ratio current	λ (nm) (A W ⁻¹) current ratio current	$(A W^{-1})$ current ratio current	current ratio current	curren	t (A)	Area (cm ²)	$Hz^{0.5}.W^{-1}$)	time	time	Type	Material
2012 280-470 1.82 1.2×10^2 $1.8 \times$	$280-470 1.82 1.2 \times 10^2 1.8 \times$	$1.82 \qquad 1.2\times10^2 1.8\times$	1.2×10^2 $1.8 \times$	$1.8 \times$	10^{-9}	$7.85 imes10^{-3}$	$6.72 imes 10^{12}$	35	48	PC	In2O3-milled nanorods
$2012 250-500 11 \qquad 80 \qquad 1.0 \times$	250-500 11 80 $1.0 \times$	11 80 $1.0 \times$	80 1.0 imes	1.0 imes	10^{6}	$6.24 imes10^{-3}$	$1.54 imes 10^{12}$	500	1,600	MSM	In2O3 PVA-coated
											nanoparticles
2003 – – 1×10^3 ~ $3.8 \times$	1×10^3 $\sim 3.8 \times$	- 1 × 10 ³ ~3.8 ×	1×10^3 $\sim 3.8 \times$	$\sim 3.8 \times$	10^{-8}	${\sim}3.8 imes 10^{-11}$	I	I	I	FET	SnO2 nanowires
$2007 1 \times 10^5 4.0 \times 10^{10}$	1×10^5 4.0×10^{10}	- 1×10^5 4.0×10^{10}	1×10^5 4.0×10^{-10}	4.0×10	$)^{-10}$	I	I	I	I	FET	SnO ₂ nanobelts
$2012 210-630 - 3.5 6.55 \times 1$	$210-630 - 3.5 6.55 \times 1$	$-$ 3.5 6.55 \times 1	$3.5 \qquad 6.55 \times 1$	6.55×1	10^{-5}	Ι	Ι	>50 s	>50 s	PC	SnO ₂ nanonets
$2008 10^5 3.2 \times 10^5$	$ 10^5$ 3.2×10^{-10}	- 10 ⁵ 3.2 × 10 ⁻	10^5 3.2 × 10 ⁻	$3.2 \times 10^{\circ}$	8-	Ι	Ι	I	I	PC	SnO2 nanowires
$2006 254 - 6.67 \times 10^2 1.5 \times 10$	$254 - 6.67 \times 10^2 1.5 \times 10$	- 6.67×10^2 1.5×10	6.67×10^2 1.5×10	1.5 imes 10	-11	I	Ι	0.22	0.09	PC	β -Ga ₂ O ₃ nanowires
$2010 254 - 3.0 \times 10^4 2.0 \times 10^{-10}$	254 – 3.0×10^4 2.0×10^1	- 3.0×10^4 2.0×10^1	3.0×10^4 2.0×10^7	2.0×10^{-10}	-13	I	I	I	< 0.02	PC	Bridged β -Ga ₂ O ₃
											nanowires
$2007 254 - 40 2.50 \times 10$	$254 - 40 2.50 \times 10$	- 40 2.50×10	$40 \qquad 2.50 \times 10$	2.50×10	-10	Ι	Ι	30	300	PC	CeO ₂ nanowires
2011 375	375	1	I	I		I	I	3	20	FET	WO ₃ nanowires
$2010 280-320 - 1.5 \times 10^2 1.00 \times 10^{-10}$	$280-320 - 1.5 \times 10^2 1.00 \times 10^2$	- 1.5×10^2 1.00×10	1.5×10^2 1.00×10	1.00×10	$)^{-10}$	I	I	100	150	PC	WO ₃ nanowire

Table 36.4 Performance comparison of In_2O_3 , SnO_2 , β -Ga₂O₃, CeO₂, and WO₃

photon coupled to the electron oscillations results in an absorption band. This is known as surface plasmon resonance (SPR). Figure 36.3 demonstrates this phenomenon in detail [62]. Polarization within the metal sphere is induced by the incoming light where the separation of free electrons with respect to the ionic core results in the need for a force to restore the system. The dipolar oscillation of electrons is the surface plasmon oscillation. The size, shape, distance between nanostructures, and material all have an effect on the resonant wavelength. The electromagnetic interaction with metal gratings was first documented in 1902 by R. Wood [63]. The Mie Theory, developed in 1908, begins to create a working model for the optical properties that result with isolated metal nanoparticles. Specifically. it is a mathematical description of the scattering of electromagnetic reaction by spherical particles in a continuous medium [64]. Fano et al. first described plasmons from incident electromagnetic radiation on metallic gratings as "superficial waves traveling with momentum along the surface of the grating and exponentially damped in the normal direction" [65]. In the review article by Ghosh et al., the assumptions, application, and limitations of the Mie Theory for plasmonic systems are thoroughly analyzed [64]. For isolated nanoparticles less than 100 nm in diameter, the Mie Theory can be reduced to the following relationship:

$$\sigma_{ext} = 9 \frac{\omega}{c} \epsilon_m^{\frac{3}{2}} V \frac{\epsilon_2(\omega)}{\left[\epsilon_1(\omega) + 2\epsilon_m\right]^2 + \left[\epsilon_2(\omega)\right]^2}$$

where $V = (4\pi/3)R^3$ is the volume of the spherical particle, ω is the angular frequency of the exciting light, c is the velocity of light, and ϵ_m and $\epsilon_1(\omega) + i\epsilon_2(\omega)$ are the dielectric functions of the surrounding medium and the material itself, respectively. The resonance condition is fulfilled when $\epsilon_1(\omega) = -2\epsilon_m$ if ϵ_2 is small or weakly dependent on ω . Nanoparticles larger than 100 nm require a correction factor that results in the broadening and red-shifting of the resonance frequency. Non-spherical nanoparticles also require a correction factor to more accurately predict the resonant frequency.

There is an ongoing debate about the exact mechanism responsible for enhancement in grating structures. Crouse and Keshavareddy weighed the significance of electromagnetic modes responsible for transmittance peaks including horizontally oriented surface plasmons (HSPs), Wood-Rayleigh (WR) anomalies, vertical surface resonances (VSRs), diffracted modes, and cavity modes (also called waveguide modes) [66]. Most agree that HSPs should be minimized as it has a converse affect on enhancement.

Ebbessen et al. were the first to discover enhanced optical transmission from sub-wavelength hole arrays [67]. This work spawned studies of 1D and 2D nanoarrays with slits, gratings, resonant cavities, and holes for the performance improvement of photodetectors [66, 68–76]. Collin et al. investigated the strong confinement of light by a resonant cavity on an MSM photodetector using



Fig. 36.4 Examples of metal nanostructures to induce surface plasmon resonance. (a) Alternating metal–semiconductor gratings on a Bragg reflector (Modified from [74]). (b) Graphene photodetector enhanced by gold nanoparticle arrays (Modified from [77]). (c) Nanoslit metal grating (Modified from [79]). (d) Aluminum nanoparticles on GaN photodetector (Modified from [78])

alternating metal–semiconductor gratings on a multilayer Bragg reflector [74]. A graphene photodetector was enhanced by gold nanoparticle arrays [77]. Plasmonic enhancement in the UV range was achieved by Butan et al. using aluminum nanoparticles deposited on a GaN MSM photodetector [78]. Schematic designs of the enhancement structures are shown in Fig. 36.4. Table 36.5 is a cross section of the experimental research results for plasmonic enhancement for various photodetectors.

			Material (semiconductor/metal	Device		
Ref.	Year	Plasmonic structure	plasmonic)	structure	λ (nm)	Enhancement factor
[76]	1998	Island nanoparticles	Si/Au	SOI-MSM	Near-IR (800 nm)	18× photocurrent enhancement
[80]	2002	Metallic gratings	Ge/Au	MSM	IR	8× absorption enhancement factor
[74]	2004	Metal-semiconductor grating	GaAs/alternating Ag and GaAs	MSM	IR	$9 \times \text{enhancement of external quantum efficiency}$
[77]	2011	Nanoparticle array	Graphene/Au	MGM*	Green-Red (530–633 nm)	Order of magnitude improvement in QE >800 % increase in photocurrent
[81]	2006	Nanoslit(s)	HgCdTe/Au	1	Mid-IR (10 µm)	 1 slit 10× enhancement factor 20 periods 250× enhancement factor
[82]	2005	Nanoparticles	Si/Au	PN junction diode	Red (640 nm)	50-80 % increase in photocurrent
[78]	2012	Nanoparticle array	GaN/AI	MSM	UV (340 nm)	50 % enhancement in QE
*Aver	age photc	current enhancement				

Table 36.5 Surface plasmon-enhanced photodetectors across UV-vis-IR spectrum



Fig. 36.5 Transient response of photodetector with MWCNT (*red*) and without MWCNT (*blue*) (b) Energy band diagram of Al/ZnO NWs/MWCNT system. (c) Carrier transport process of ZnO NWs/MWCNT interface (Republished with permission from [98])

Carrier Transfer Mechanisms

It is well known that the photoresponse of metal oxide materials is sensitive to film thickness, deposition temperature, annealing treatments, and doping. It has been shown that aspects of device performance can therefore be improved by optimizing these parameters [83–88]. Further enhancement with the addition of metal/semimetal layers is an approach to circumvent issues inherent to most metal oxides attributed to the slow chemisorption of oxygen molecules [16].

In 2007, Yadav et al. were the first to study the interfacial charge-transfer kinetics between ultrathin layers of different metals with a metal oxide thin film [89]. In this work, enhanced photoresponse was due to the effective transfer of electrons from the metal to ZnO at the interface, and the injected charge carriers compensate the surface states and increase the photoconductivity. The UV/dark current ratio improved by more than $5 \times$ with Te/ZnO bilayer. Nevertheless, the rise and fall times significantly increased, demonstrating the common trade-off in photodetectors. Tzeng et al. continued the concept of interfacial charge transfer by modifying the interface using Ag nanoparticles. The photocurrent-to-dark-current contrast ratio increased by almost 74 times. The decay time was reduced from 824 to 36 s, and the rise time was also reduced from ~151 to 4 s after depositing Ag NPs [90]. These results are

Ref.	Year	Carrier transfer enhancement material	Semiconductor material	Device structure	λ (nm)	Enhancement factor
[<mark>92</mark>]	2010	Embedded graphene layer	ZnO QD-polymer		365 nm	\sim 330× photocurrent
[89]	2007	Metal thin sheet: Te, Au, Pb, Cu, Sn, Al	ZnO thin film	Bilayer	365 nm	5× UV/dark current ratio (Te)
[<mark>90</mark>]	2012	Ag NP	ZnO NW		365 nm	$74 \times$ UV/dark current ratio
						$37 \times$ faster rise time
						$23 \times$ faster fall time
[101]	2013	Graphene core- shell	ZnO NP	PC	335 nm	3 orders of magnitude rise and fall time improvement
[98]	2013	MWCNT	ZnO NW/p-Si	PD	335 nm	3× improvement responsivity
						$23 \times$ faster rise time
						$18 \times$ faster fall time
[<mark>94</mark>]	2013	Graphene core– shell	WO ₃	MSM	340 nm	100× faster time response
[<mark>99</mark>]	2012	Graphene QD	ZnO NP	MSM	335 nm	2× improvement responsivity
						$50 \times$ faster rise time
						$37 \times$ faster fall time

Table 36.6 Carrier transfer enhancement of UV photodetectors

attributed to the formation of AgO_x at the interface between Ag and ZnO, which increased the Schottky barrier height and led to a significant enhancement in the charge-transfer kinetics near the interface of ZnO and AgO_x .

Carbon nanomaterials, specifically carbon nanotubes (CNTs) and graphene, have been investigated in hybrid nanocomposites to provide improved charge separation and subsequent carrier transport [91–97]. CNT hybrid materials have proven to be a versatile electrode for fuel-cell, photodetector, and supercapacitor applications for their solution processable fabrication, high conductivity, mechanical flexibility and strength, and dimension variability. Recently, Shao et al. demonstrated a multiwalled carbon nanotube (MWCNT)/ZnO NW/p-Si photodiode with rise and fall times of 0.09 s and 0.08 s, respectively. This is at least 20 times faster than ZnO NW/p-Si photodiode without using MWCNTs. In addition to improved transient response, the maximum responsivity of the photodiode employing MWCNTs enhanced by three times, which can reach to 4.7 A/W for UV light (365 nm) at 2 V bias [98]. The improved performance is explained by the band diagram in Fig. 36.5. Electrons transfer from ZnO to MWCNTs because it is energetically favorable to do so. Due to the semimetallic property of the MWCNTs, the composite structure has much better conductivity than the bare ZnO NWs array. This effectively improved the carrier transport and collection efficiency.

Graphene has many unique properties including high carrier transport mobility, superior mechanical properties, excellent chemical stability, larger surface area than CNTs, and low cost. Thin graphene sheet layers, quantum dots, and core-shells are common structures for nanocomposites with metal oxides for improved carrier transport [93, 94, 97, 99]. Graphene-CdS quantum dot nanocomposites were explored through facile fabrication processes by Wu et al. and Cao et al. as methods to enhance conductivity of quantum dot layers. Williams et al. were among the first to create graphene metal oxide nanocomposites for conductivity improvement [100]. Recently, graphene quantum dots, synthesized by the hydrothermal method, were used to enhance the performance of ZnO-based ultraviolet photodetectors through the preferential carrier transfer to the more conductive material [99]. The rise and fall times reduced from longer than 3 s each to 0.06 s and 0.08 s, respectively, while a high responsivity of >450 A/W at 375 nm was maintained [99]. Similar results were obtained with a ZnO NPs-graphene core-shell structure [94, 101] (Table 36.6).

Summary

Metal oxides and their nanostructures have many advantages for ultraviolet detection including low cost, a high surface-to-volume ratio, and a large internal gain mechanism. A long transient response due to the adsorption of oxygen on the semiconductor surface limits performance. Metal and semimetal nanostructures can improve overall performance by inducing surface plasmon resonance or allowing the transfer of carriers to more conductive material.

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